

EX NAVODAYAN FOUNDATION

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Full Syllabus

JEE-Main

Paper-4

Test Date:

M.M:300

TEST INSTRUCTIONS

- 1. The test is of **3 hours** duration.
- 2. The test booklet consists of **75 questions**.
- 3. The maximum marks are **300**.
- 4. All questions are compulsory.
- 5. There are three parts in the questions paper consisting of Physics, Chemistry and Mathematics having **25** questions in each part.

Each Parts Contains –

- 20 multiple choice questions. Each question has four choices (a), (b), (c) and (d) out of which ONLY
 ONE is correct. All questions are carrying +4 marks for right answer and -1 mark for wrong answer.
- 05 questions with answer as **numerical value** all questions are carrying **+4 marks** for right answer and **-1 marks** for wrong answers.

Name of the Candidate (in Capital Letter): ______

Registration No. _____

Invigilator Signature

Physics

(Single Correct Choice Type)

This Section contains **20 multiple choice questions.** Each question has four choices (a), (b), (c) and (d) out of which **ONLY ONE** is correct.

- A boy is cycling at 20 km/hr in a direction making on angle 30° north of east. Find the velocity of a second boy cycling towards north so that to him first boy appears to be moving towards east.
 (a) 5 km/hr
 (b) 10 km/hr
 (c) 20 km/h
 (d) 15 km/hr
- 2. The potential energy of a particle of mass 1 kg in a force field is given as $U(x) = \left(\frac{x^4}{4} \frac{x^2}{2}\right)J$. The

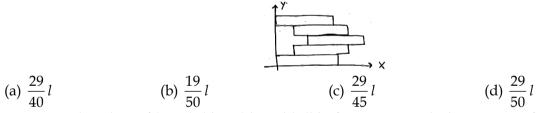
total mechanical energy of the particle is 2J then find the maximum speed of particle in its motion.

(a)
$$\sqrt{3} \text{ m/s}$$
 (b) $\frac{3}{\sqrt{2}} \text{ m/s}$ (c) $\frac{1}{\sqrt{2}} \text{ m/s}$ (d) $\frac{3}{2} \text{ m/s}$

3. When road of a circular turn is dry the maximum remissible speed for taking a turn on it is 10 m/s. Find the maximum turning speed on wet road when the frictional coefficient between road and tyres reduces to half.

(a)
$$5\sqrt{2}$$
 m/s (b) $\frac{5}{\sqrt{2}}$ m/s (c) 10 m/s (d) $\frac{5}{2}$ m/s

 Figure shows five uniform brick each of length *l*. Each brick is placed over another with an offset of length *l*/10. Find the x co-ordinate of center of mass of the system.

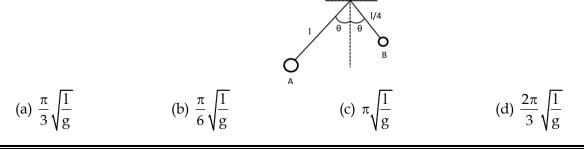


- 5. A rectangular plate of length '*l*' and breadth 'b' of mass m. Find it's moment of inertia about an axis along its diagonal.
 - (a) $\frac{1}{6} \frac{ml^2 b^2}{(l^2 + b^2)}$ (b) $\frac{1}{3} \frac{ml^2 b^2}{(l^2 + b^2)}$ (c) $\frac{ml^2 b^2}{(2l^2 + b^2)}$ (d) $\frac{ml^2 b^2}{2(l^2 + b^2)}$

6. The ratio of thermal conductivities of two materials of different rods of same cross-section is 5 : 3.If the thermal resistance of there rods are in ratio 1 : 3. Find the ratio of length of there rods.

(a)
$$\frac{4}{3}$$
 (b) $\frac{5}{1}$ (c) $\frac{5}{9}$ (d) $\frac{9}{5}$

7. Figure shows two pendulums A and B of length l' and l/4' released from rest from the position shown in figure. Calculate the time instant when the two string become parallel for first time.



8.	Sound from two id	lentical sources S_1 and	S ₂ reaches a point (p)	where intensity of sound is							
	observed to be I_0 . Point p is equidistant from there sources. If intensity of S_1 is increased to 3 time										
	and that of S_2 is decr	eased by 3 times. Find t	he sound intensity at poi	int (p).							
	(a) $\frac{I_0}{6}$	(b) $\frac{16}{9}I_0$	(c) $\frac{16}{3}I_0$	(d) $\frac{16}{11}I_0$							
Э.	An air bubble of radius 1 cm rising at a steady rate of 0.5 cm/s through a liquid of density 0.8										
	gm/cm ³ . Calculate the coefficient of viscosity of the liquid. {Neglect the mass of air}										
	(a) 32 poise	(b) 38.5 poise	(c) 35.6 poise	(d) None of these							
0.	A body floats in water with 40% of it's volume outside water. When the same body floats in of										
	60% of its volume remain outside oil. Find the specific gravity of oil.										
	(a) 2	(b) 3.5	(c) 1.8	(d) 1.5							
11.	Due to some reasons	s if earth radius decrease	e by 6% with it's mass re	maining unchanged, what wil							
	happen to the acceleration due to a gravity on surface of earth.										
	(a) will decrease by	9.6%	(b) will increase by	e of earth. will increase by 8%							
	(c) will increase by 1	1.6%	(d) will decrease b	y 12.5%							
12.	Find the angle with dipole at which electric field due to dipole is normal to dipole moment.										
	(a) $\theta = \tan^{-1}(\sqrt{3})$	(b) $\theta = \tan^{-1}(2)$	(c) $\theta = \tan^{-1}(3)$	(d) $\theta = \tan^{-1}(\sqrt{2})$							
3.	Two electric charge	s 'q' and '–2q' are place	ced at (0, 0) and (6, 0)	points, find the locus of zero							
		Two electric charges 'q' and ' $-2q'$ are placed at (0, 0) and (6, 0) points, find the locus of zero pointential points in the co-ordinate system.									
	(a) parabola	(b) ellipse	(c) circle	(d) line							
14.	Figure shows four	parallel plates with son	ne connections. If area c	of plates is 'A' and separatior							
	-		ivalent capacitance acros								
	, 01	r F									
			X								
	(a) $\frac{2\varepsilon_0 A}{d}$	(b) $\frac{2\varepsilon_0 A}{3d}$	(c) $\frac{\varepsilon_0 A}{2d}$	(d) $\frac{\varepsilon_0 A}{3d}$							
15.	u		lowing in 10Ω resistance	34							
		1	0V 402								
		20.A	102								
	/ N = /		207								
16.	(a) 5A Two insulated circu	(b) 2A lar loop 'A' and 'B' of	(c) 4A radius 'a' carrying a cu	(d) 0A rrent 'I' in the anti clock wise							
			e of the magnetic induction								
		e									
			I								
	$\sqrt{\sqrt{2}\mu_0}$ I	(b) $\frac{\mu_0 I}{\sqrt{2}a}$	(c) $\frac{\mu_0 I}{2a}$	(d) $\frac{2\mu_0 I}{r}$							
	(a) <u> </u>	(b) <u></u>	$(C) \xrightarrow{\cdot \cdot \cdot \cdot}$	(d) <u> </u>							

17. A long conducting wire having a current 'I' flowing through it, is bent into a circular coil of 'N' turns. Then it is bent into a circular coil of 'n' turns. The magnetic field is calculated at the centre of coils in both the cases. The ratio of magnetic field in first case to that of second case is

(a)
$$N^2: n^2$$
 (b) $N: n$ (c) $n: N$ (d) $n^2: N^2$

A wire of length 1m moving with velocity 8 m/s at right angles to a magnetic field of 2T. The magnitude of induced emf, between the end's of wire will be

19. A proton, an electron and an alpha particle have the same energies, their de-broglie wavelength will be compared as

$$(a) \lambda_{p} > \lambda_{e} > \lambda_{\alpha} \qquad (b) \lambda_{\alpha} > \lambda_{p} > \lambda_{e} \qquad (c) \lambda_{e} > \lambda_{\alpha} > \lambda_{p} \qquad (d) \lambda_{p} < \lambda_{e} < \lambda_{\alpha}$$

20. A lens is made up of 3 different transparent media as shown in figure. A point object O is placed on its axis beyond 2f. How many real images will be obtained on the other side.

(a) 2	(b) 1
(c) No image will be found	(d) 3

(Integer Type Questions)

This Section contains **05** Questions. The answer to each question is a single digit integer ranging from 0 to 9. The correct digit below the question number in the ORS is to be bubbled.

21. ϵ_0 and μ_0 are the electric permittivity and magnetic permeability of free space respectively. If the corresponding quantities of a medium are $2\epsilon_0$ and $1.5 \mu_0$ respectively, the refractive index of medium is \sqrt{x} then value of 'x' is

- 22. A silver wire has mass (0.6 ± 0.006) g and radius (0.5 ± 0.005) mm and length (4 ± 0.04) cm. The maximum percentage error in the measurement of it's density will be
- 23. A body is launched up an inclined plane with inclination ' α ' to horizontal. It is observe that the coeff of friction is $\frac{3}{5}$ tan α then ratio of time of decent and acent along length of inclined is
- 24. A small block slides down an inclined plane with inclination angle ' α ' with horizontal. The coefficient of friction depends on the distance 'x' covered as $\mu = kx$, where 'k' is constant distance travelled by block till it stops is $\frac{n \tan \alpha}{k}$ then value of 'n' is
- 25. Wall of a container can beer a maximum pressure of 10^6 Pa. It encloses a gas at a pressure of 8×10^5 Pa at 300 K temperature. If temperature of container is gradually increased, find the temperature at which container will break. (in kelvin)

Chemistry

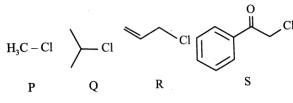
(Single Correct Choice Type)

1.	Consider the following reaction:											
	$xMnO_4^- + yC_2O_4^{2-} + zH^+ \rightarrow xMn^{2+} + 2yCO_2 + \frac{z}{2}H_2O$											
	The value's of x, y and z in the reaction are, respectively:											
	(a) 5, 2 and 16 (b) 2, 5 and 8	(c) 2, 5 and 16	(d) 5, 2 and 8									
2.	Experimentally it was found that a metal oxide has formula $M_{0.98}O$. Metal M, present as M^{2+} and											
	M ³⁺ in its oxide. Fraction of the metal which exists as M ³⁺ would be:											
	(a) 7.01% (b) 4.08%	(c) 6.05%	(d) 5.08%									
3.	The electrons identified by quantum numbers n and l :											
	(A) $n = 4, l = 1$ (B) $n = 4, l = 0$	(C) $n = 3, l = 2$	(D) n = 3, $l = 1$									
	can be placed in order of increasing energy as	:										
	(a) $C < D < B < A$ (b) $D < B < C < A$	(c) $B < D < A < C$	(d) $A < C < B < D$									
4.	For which of the following molecule significar	t $\mu \neq 0$?										
	$\begin{array}{ccc} & & & & & \\ & & & \\ Cl & & CN & & OH & SH \\ (a) \text{ Only i} & & (b) \text{ i and ii} \end{array}$	(c) Only iii	(d) iii and iv									
5.	A piston filled with 0.04 mol of an ideal ga	s expands reversibly from	m 50.0 mL to 375 mL at a									
	constant temperature of 37.0°C. As it does so, it absorbs 208 J of heat. The values of q and w for the											
	process will be: (R = 8.314 J/mol K) (ln 7.5 = 2.01)											
	(a) $q = +208 J$, $w = -208 J$	(b) q = -208 J, w = -2	208 J									
	(c) $q = -208 J$, $w = +208 J$	(d) q = +208 J, w = +	208 J									
6.	The following reaction is performed at 298 K.											
	$2NO(g) + O_2(g) \rightleftharpoons 2NO_2(g)$											
	The standard free energy of formation of NO(g) is 86.6 kj/mol at 298 K. What is the standard free											
	energy of formation of NO ₂ (g) at 298 K? ($K_p = 1.6 \times 10^{12}$)											
	(a) $86600 - \frac{\ln(1.6 \times 10^{12})}{R(298)}$	(b) 0.5[2 × 86,600 – R	2(298) $\ln(1.6 \times 10^{12})$]									
	(c) R(298) $\ln(1.6 \times 10^{12})$ – 86600	(d) 86600 + R(298) lr	$n(1.6 \times 10^{12})$									
7.	The equilibrium constant at 298 K for a reaction A + B \rightleftharpoons C + D is 100. If the initial concentration of all the four species were 1 M each, then equilibrium concentration of D in mol L ⁻¹ will be: (a) 1.818 (b) 1.182 (c) 0.182 (d) 0.818											

IIT

(a)

8. KI in acetone, undergoes S_N 2 reaction with each of P, Q, R and S. The rates of the reaction vary as



(b) none of these

$$P > Q > R > S$$
 (b) $S > P > R > Q$ (c) $P > R > Q > S$ (d) $R > P > S > Q$

9. The IUPAC name(s) of the following compound is(are)

(a) 1-chloro-4-methylbenzene

- (c) 4-methylchlorobenzene (d) 1-methyl-4-chlorobenzene
- 10. Which of the following compounds will exhibit geometrical isomerism?
 - (a) 2-Phenyl-1-butene (b) 1, 1-Diphenyl-1-propene
 - (c) 1-Phenyl-2-butene (d) 3-Phenyl-1-butene

Liquid 'M' and liquid 'N' form an ideal solution. The vapour pressures of pure liquids 'M' and 'N' are 450 and 700 mmHg, respectively, at the same temperature. Then correct statement is:
(x_M = Mole fraction of 'M' in solution; x_N = Mole fraction of 'N' in solution;
Y_M = Mole fraction of 'M' in vapour phase; y_N = Mole fraction of 'N' in vapour phase)

(a) $\frac{x_{\rm M}}{x_{\rm N}} = \frac{y_{\rm M}}{y_{\rm N}}$ (b) $(x_{\rm M} - y_{\rm M}) < (x_{\rm N} - y_{\rm N})$

(c)
$$\frac{x_{\rm M}}{x_{\rm N}} < \frac{y_{\rm M}}{y_{\rm N}}$$
 (d) $\frac{x_{\rm M}}{x_{\rm N}} > \frac{y_{\rm M}}{y_{\rm N}}$

12. Resistance of 0.2 M solution of an electrolyte is 50 Ω . The specific conductance of the solution is 1.4 S m⁻¹. The resistance of 0.5 M solution of the same electrolyte is 280 Ω . The molar conductivity of 0.5 M solution of the electrolyte in S m² mol⁻¹ is:

(a)
$$5 \times 10^{-4}$$
 (b) 5×10^{-3} (c) 5×10^{3} (d) 5×10^{2}

13. The half life period of a first order chemical reaction is 6.93 minutes. The time required for the completion of 99% of the chemical reaction will be $(\log 2 = 0.301)$

(a) 23.03 minutes (b) 46.06 minutes (c) 460.6 minutes (d) 230.03 minutes

14. Among the following oxoacids, the correct decreasing order of acid strength is:

- (a) $HOCl > HClO_2 > HClO_3 > HClO_4$ (b) $HClO_4 > HOCl > HClO_2 > HClO_3$ (c) $HClO_4 > HClO_3 > HClO_2 > HOCl$ (d) $HClO_2 > HClO_4 > HClO_3 > HOCl$
- 15. Which of the following pairs represent linkage isomers?
 - (a) $[Pd(P Ph_3)_2(NCS)_2]$ and $[Pd (P Ph_3)_2 (SCN)_2]$ (b) $[Co(NH_3)_5NO_3]SO_4$ and $[Co(NH_3)_5SO_4]NO_3$
 - (c) $[PtCl_2(NH_3)_4]Br_2$ and $[PtBr_2(NH_3)_4]Cl_2$ (d) $[Cu(NH_3)_4]PtCl_4]$ and $[Pt(NH_3)_4][CuCl_4]$

16. Consider the following complex ions, P, Q and R. P = $[FeF_6]^{3-}$, Q = $[V(H_2O)_6]^{2+}$ and R = $[Fe(H_2O)_6]^{2+}$ The correct order of the complex ions, according to their spin-only magnetic moment values (in B.M.) is (a) R < Q < P(b) Q < R < P(c) R < P < Q(d) Q < P < R17. Consider the following bromides: Me Br 🧠 Me⁻ Br Br (A) **(B)** (C) The correct order of S_N1 reactivity is (a) B > C > A(b) B > A > C(c) C > B > A(d) A > B > C18. In the reaction, $CH_3COOH \xrightarrow{\text{LiAH}_4} A \xrightarrow{\text{PCl}_5} B \xrightarrow{\text{Alc.KOH}} C$, the product C is: (a) Acetaldehyde (c) Ethylene (d) Acetyl chloride (b) Acetylene 19. Arrange the following amines in the decreasing order of basicity: Η Η Π III T (a) I > II > III(b) III > I > II (c) III > II > I (d) I > III > II20. The term anomers of glucose refers to (a) enantiomers of glucose (b) isomers of glucose that differ in configuration at carbon one (C-1)

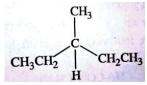
(c) isomers of glucose that differ in configurations at carbons one and four (C-1 and C-4)

(d) a mixture of (D)-glucose and (L)-glucose

(Integer Type Questions)

This Section contains **05** Questions. The answer to each question is a single digit integer ranging from 0 to 9. The correct digit below the question number in the ORS is to be bubbled.

- 21. Total number of isomers, considering both structural and stereoisomers of cyclic ethers with the molecular formula C_4H_8O is _____
- 22. In an atom, the total number of electrons having quantum numbers n = 4, $|m_1| = 1$ and $m_s = -\frac{1}{2}$ is
- 23. Among B₂H₆, B₃N₃H₆, N₂O, N₂O₄, H₂S₂O₃, H₂S₂O₈, the total number of molecules containing covalent bond between two atoms of the same kind is _____
- 24. In 1 L saturated solution of AgCl $[K_{sp}(AgCl) = 1.6 \times 10^{-10}]$, 0.1 mol of CuCl $[K_{sp}(CuCl) = 1.0 \times 10^{-6}]$ is added. The resultant concentration of Ag⁺ in the solution is 1.6×10^{-x} . The value of "x" is
- 25. The maximum number of isomers (including stereoisomers) that are possible on monochlorination of the following compound is



Mathematics

(Single Correct Choice Type)

This Section contains **20 multiple choice questions.** Each question has four choices (a), (b), (c) and (d) out of which **ONLY ONE** is correct.

1.
$$\int_{0}^{\frac{\pi}{2}} \sin^{20} x \cdot \cos^{20} x \, dx = \frac{1}{\lambda} \int_{0}^{\frac{\pi}{2}} \sin^{20} x \, dx, \text{ then } \lambda \text{ is equal to}$$
(a) 2^{20} (b) 2^{19} (c) 2^{-20} (d) 2^{19}
2. If A is 3×3 matrix such that $A^{T} = 5A + 2I$, where A^{T} is the transpose of A and I is the 3×3 . Identify matrix, then there exist a column matrix $x = \begin{bmatrix} x \\ y \\ z \end{bmatrix} \neq \begin{bmatrix} 0 \\ 0 \\ 0 \end{bmatrix}$. Then AX is equal to
(a) $AX = X$ (b) $AX = -\frac{X}{2}$ (c) $AX = -2X$ (d) $AX = 0$
3.
$$\lim_{n \to \infty} \sum_{r=1}^{2n} \frac{r}{\sqrt{n^{2} + r^{2}}}$$
 equals
(a) $\sqrt{5}$ (b) $\sqrt{2} - 1$ (c) $\sqrt{5} - 1$ (d) $1 + \sqrt{2}$
4. If f(x) is a polynomial function satisfying $f(x) \cdot f(\frac{1}{x}) + 3f(x) + 3f(\frac{1}{x}) = 0$ and $f(3) = 24$, then the value of $f(-2) + f(2)$ is
(a) 2 (b) -6 (c) 0 (d) None of these
5. The locus of the midpoint of the chord of the ellipse $49x^{2} + 16y^{2} = 784$. The tangent at ends of which intersect on the circle $x^{2} + y^{2} = 100$ is
(a) $(49x^{2} + 16y^{2})^{2} = (\frac{784}{10})^{2} (x^{2} - y^{2})$ (b) $49x^{2} + 16y^{2} = \frac{784}{10}$
(c) $(49x^{2} + 16y^{2})^{2} = (\frac{784}{10})^{2} (x^{2} + y^{2})$ (d) None of these
6. The maximum value of $y = 4 \cos 2x + 3 \sin x + 5 \text{ is equal to}$
(a) 10 (b) $\frac{297}{32}$ (c) 0 (d) None of these
7. The sum $\sum_{r=0}^{20} xC_{r}$, $\sin(rx)$, $\cos(30 - r)x$ is equal to
(a) $2^{29} \cos 30 x$ (b) $2^{29} \cos 30 x$ (c) $2^{29} \sin 29 x$ (d) $2^{29} \sin 30 x$
8. If a, b, c \in are distinct number in A.P a, α , b are in G.P, b, β , c are also in G.P, then α^{2} , b^{2} , β^{2} will be in
(a) $A.P$ (b) G.P (c) H.P (d) None of these

IIT If the plane passing through the points (a, 1, 1) (1, 2, 1) and (2, 3, 4) is parallel to the line 9. $\vec{r} = \lambda(\hat{i} - \hat{j} + 2\hat{k})(\lambda \in R)$, then a is equal to (c) $\frac{3}{2}$ (a) $-\frac{1}{2}$ (b) -1 (d) 0The area $s \cap s'$ is where $S = \left\{ (x, y); \frac{y(3x-1)}{x(3x-2)} < 0 \right\}$ and 10. $S' = \{(x, y) \in A \times B, -1 \le A \le 1 \text{ and } -1 \le B \le 1\}$ (a) 1 (b) 2(c) 3 (d) 4 Let chord of contact to drawn from Every point lying on circle $x^2 + y^2 = 36$ to the ellipse 11. $\frac{x^2}{4} + \frac{y^2}{9} = 1$ such that all the lines touches an standard ellipse whose electricity is (a) $\frac{\sqrt{65}}{0}$ (b) $\frac{\sqrt{5}}{2}$ (c) $\frac{4}{5}$ (d) None of these 12. Let tangents PA & PB are drawn from variable point P on the parabola $y^2 = 4x$ to the circle $x^2 + y^2 + y^2$ 6x - 4y - 3 = 0 then the director circle of locus of circumcenter of $\triangle PAB$ is (b) $x^2 + y^2 + 6x - 4y - 19 = 0$ (c) $x = -\frac{1}{2}$ (a) x + 2 = 0(d) None of these The differential equation of the curve $\frac{x}{a-1} + \frac{y}{a+1} = 1$ is given by 13. (a) (v' - 1)(v + xv') = 2v'(b) (y' + 1)(y + xy') = y'(c) (y' + 1)(y - xy') = 2y'(d) None of these $cosgn(cos^{-1} x - cos^{-1} x^2), x > 0$ If $f(x) = \begin{cases} \frac{\sin^{-1} x - x}{x^3}, & x < 0 \text{ is continuous at } x = 0, \text{ then a + b equal to} \\ x = 0 \end{cases}$ 14. (b) $\frac{1}{2}$ (c) $-\frac{1}{2}$ (a) 0 (d) None of these 15. From first 100 natural number, 3 numbers are selected if these numbers are in A.P. then find probability that these numbers are even. (c) $\frac{29}{49}$ (b) $\frac{29}{66}$ (d) $\frac{12}{49}$ (a) $\frac{1}{66}$ $\int \frac{3\cos x}{2\cos x + 5\sin x} dx$ is equal to 16.

(b) $\frac{6}{29}x - \frac{15}{29}\ln|2\cos x + 5\sin x| + C$ (a) $\frac{15}{29}x + \frac{6}{29}\ln|2\cos x + 5\sin x| + C$ (c) $\frac{6}{29}x + \frac{15}{29}\ln|2\cos x + 5\sin x| + C$ (d) None of these

[Note: where C is integration constant]

17. Let
$$S_n = \cot^{-1}\left(6x + \frac{2}{x}\right) + \cot^{-1}\left(10x + \frac{2}{x}\right) + \cot^{-1}\left(15x + \frac{2}{x}\right) + \dots + n \text{ term, where } x > 0$$

If $\lim_{n \to \infty} S_n = 1$ then x equals
(a) $\cot 1$ (b) $\frac{2}{3} \cot 1$ (c) $\frac{3}{2} \tan 1$ (d) None of these
18. If x, y, z satisfy the system of equation $\tan^2 x + \cot^2 x = 2 \cos^2 y$ and $\cos^2 y + \sin^2 z = 1$
then value of $\int_{\cos^2 y}^{\sec^2 x + \sec^2 z} \frac{t^2}{t^2 - 4t + 8} dt$ is equal to
(a) 0 (b) 1 (c) 2 (d) 3
19. If $\arg\left(\frac{z_1 - \frac{z}{|z|}}{\frac{z}{|z|}}\right) = \frac{\pi}{2}$ and $\left|\frac{z}{|z|} - z_1\right| = 3$, then $|z_1|$ is equal to
(a) $\sqrt{3}$ (b) $\sqrt{26}$ (c) $\sqrt{10}$ (d) $2\sqrt{2}$
20. Let \vec{r} be a position vector of a variable point in x-y plane such that $\vec{r}.(6\hat{j} - 4\hat{i} + \vec{r}) = 3$, then the maximum value of $|\vec{r} + 2\hat{i} - 3\hat{j}|$ is equal to
(a) $2 + \sqrt{13}$ (b) $2(2 + \sqrt{13})$ (c) $2 + 2\sqrt{13}$ (d) $4 + \sqrt{13}$

(Integer Type Questions)

This Section contains **05** Questions. The answer to each question is a single digit integer ranging from 0 to 9. The correct digit below the question number in the ORS is to be bubbled.

21. If
$$(1 + x + x^2 + x^3)^{100} = \sum_{r=0}^{300} a_r \cdot x^r$$
 and $\sum_{r=0}^{300} a_r = k$, then the value of $\sum_{r=0}^{300} r \cdot a_r = \lambda k$. Find value of λ
22. If $f(x) = \sin x$ and $g(x) = \underbrace{f(f(f \dots f(x)))}_{2018 \text{ times}}$, then the value of $g'(0) + g''(0) + g'''(0)$ is equal to

23. A 10 digit number start with 2 and all its digits are prime, then the probability that the sum of are two consecutive digits of the number if prime is $\frac{1}{2^k}$. Find value of k

24. Let
$$f(x) = \begin{cases} \left(\sin \frac{2x^2}{a} + \cos \frac{3x}{b} \right)^{\frac{ab}{x^2}}, & x \neq 0 \\ e^{2x+3}, & x = 0 \end{cases}$$

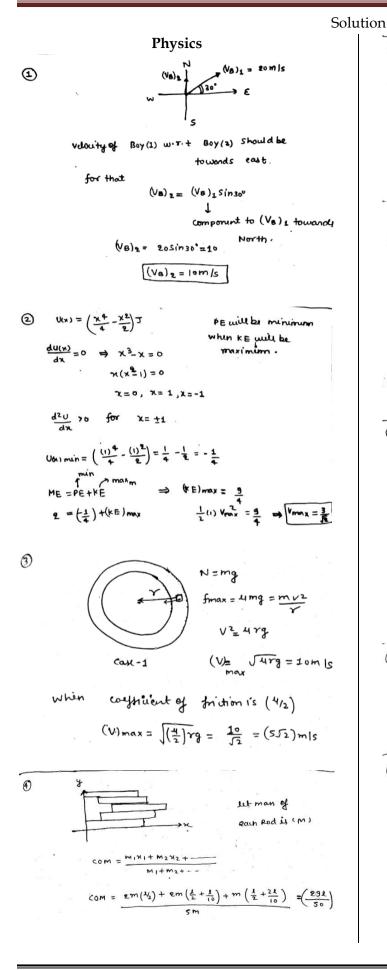
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is continuous function at x = 0, $\forall b \in R$ then $\left|\frac{1}{a_{\min}}\right|$ is equal to

25. Let α and β are the roots of the equation $x^2 - 6x + 12 = 0$. If the value of $(\alpha - 2)^{12} + \frac{(\beta - 6)^{12}}{\alpha^{12}} - 1$ is

 a^{b} , then minimum value of (a + b) is equal to

						A	nsw	er –	key						
Physic		11.	С	21.	3	6.	b	16.	b	Math	L	11.	а	21.	150
1.	В	12.	D	22.	4	7.	а	17.	а	1.	а	12.	а	22.	-2017
2.	В	13.	С	23.	2	8.	b	18.	с	2.	b	13.	с	23.	13
3.	А	14.	_	24.	2	9.	а	19.	b	3.	С	14.	а	24.	4
4.	D	15.	D	25.	375	10.	с	20.	b	4.	b	15.	d	25.	10
5.	А	16.	В	Cher	nistry	11.	d	21.	10	5.	С	16.	с		
6.	С	17.	А	1.	c	12.	а	22.	6	6.	b	17.	b		
7.	А	18.	С	2.	b	13.	b	23.	4	7.	d	18.	с		
8.	С	19.	В	3.	b	14.	с	24.	7	8.	а	19.	с		
9.	С	20.	D	4.	d	15.	а	25.	8	9.	d	20.	b		
10.	D			5.	а					10.	b				



(a) MI of redergular plat cambe.
given as

$$I = 2 (MI of the triangular plate as shown)$$

$$I = 2 \left(\frac{1}{6} \left(\frac{M}{2}\right) x^{2}\right)$$

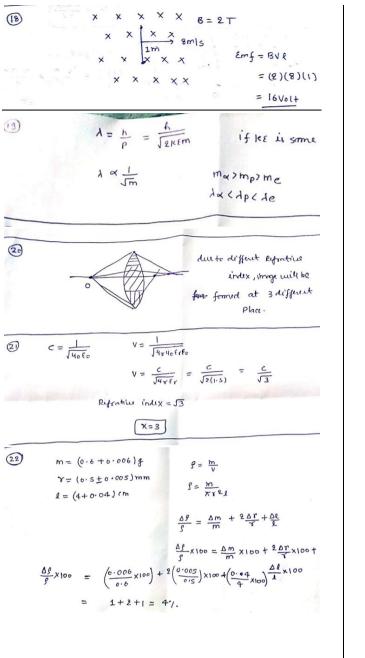
$$x = \frac{1}{6} \left(\frac{M^{2}}{2} x^{2}\right)$$

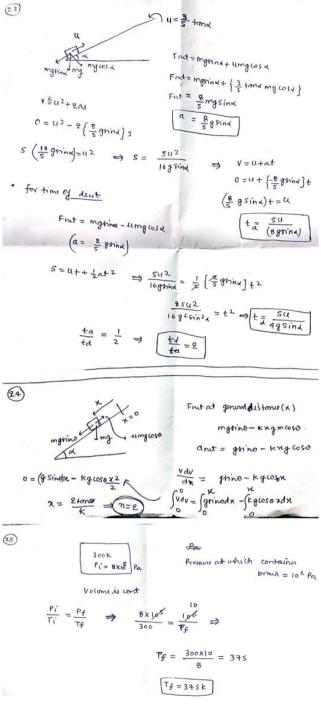
$$x = \frac{1}{7} \left(\frac{M^{2}}{2} x^{2}\right)$$

$$x = \frac{1}$$

(*) Let volume of body if (v)
for floatation
$$mg = 0.6 V Jug - (1)$$
 [ineac of
 $mg = 0.6 V S o g - (1)$
for eqn (1) e(i)
 $\frac{1}{2m} = \frac{3}{2} = 1.5$ specific gravity
(*) $g_{S} = \frac{GWe}{Br^{2}}$ $e_{E} \rightarrow 0.94 e_{E}$
 $\frac{3}{1mw} = \frac{GWe}{Br^{2}}$ $e_{E} \rightarrow 0.94 e_{E}$
 $\frac{GW}{Grad} = \frac{GWe}{Grad}$ $\frac{1146}{9.5}$
 $\frac{GW}{Grad} = \frac{GW}{Grad}$ $\frac{1146}{9.5}$
 $\frac{GW}{Grad} = \frac{11}{9.5}$
 $\frac{GW}{Grad} = \frac{1146}{9.5}$
 $\frac{GW}{Grad} = \frac{1146}{9.5}$
 $\frac{GW}{Grad} = \frac{11}{9.5}$
 $\frac{11}{9.5}$
 $\frac{11}{9.5}$

(13)





Chemistry

Q.1 Ans = (C) 2, 5 and 16 Q.2. (b) For 1 male of 0xide Ans = moles of M = 0.98, moles of $0^2 = 1$ Lets male of M3+= 1 .: males of M2+ = 0.98-x on balancing charge. (0.98-x) × 2 + 3x-2=0 K= 0.04 Q.3 (b) Ans+ According to bohr Burryh (n+e) 回へ(四) く(こ) <(A) Q. 4. (d) ;;;-()-s. Any > In both the malecule the defale moment not cancelling each other due to V-shape or bent structure. Then (1470) Q. 5. (a) Ans: In isothermal expansion 40=0 Q=-W ·· W= -2.303 NRT log V2 = -2.303 × 0.04 × 8.314 × 310 × log [W=-208] 5 [R= 208] Q.6: (b) Ans. 2 NO(3) + 02 (3) = 2 NO2(3) As All'= -RT lnkp · 246° - 246° = - R(298) In (1.6 ×10¹²) 2 AGNO2 - 2 × 86600 = - R(298) ln (1.6×1012) 201° = 2×8660 - R(298) In (1.6×10) AG° = 1/2×86600 - R(298) 2n(1.6×1012) = 0.5 [2×86600 - R(298) In (1.6×1012) 01.7 (a) $A + B \Longrightarrow C + D$ No. finitial mole 1 1 1 1 At equilebrium (1-9) (1-9) (1+9) (1+9) $K_{e} = \frac{(1+a)^{2}}{(1-a)^{2}} = 100$

on salving a = 0.81 ... [D] at equilibrium (1+4) = 1+0.81 = [1.81]

SZPZRZQ Q.8 (b) Ans: Q.9 (a) \$ (b) Ans 61.10: (C) Ans: 1-phonyl-2-butene C6 H5 CH2 - CH = CH - CH3 Ph- chs Ph-(43 G.11. (d) Pm = 450 mm Hg , Pn = 700 mm Hg Ans. PM = PM XM = YM PT $P_{M}^{o} = \frac{Y_{M}}{X_{M}} (P_{T})$ similarly $P_{M}^{o} = \frac{Y_{M}}{X_{N}} (P_{T})$ Given Pm K PN° $\Rightarrow \frac{y_{M}}{x_{M}} < \frac{y_{N}}{x_{N}} \qquad \left| \frac{y_{M}}{y_{N}} < \frac{x_{M}}{x_{N}} \right|$ $R = P \frac{1}{\alpha} = \frac{1}{k} \frac{x}{\alpha}$ Q. 12: (a) Any: $\frac{1}{2} = RXK = 50 \times 1.4 \times 10^{2}$ for 0.5 M Solution R= 280 12 ; K=7 $\frac{l}{a} = 50 \times l \cdot 4 \times l^{5^2} \Rightarrow R = P \frac{l}{a} = \frac{l}{K} \times \frac{l}{a}$ k= 1/2 × 50×1.4×152 = 1 x 70 x 102 = 2.5 x 103 Scm-1 Now $T_m = \frac{K \times 1000}{M} \Rightarrow \frac{2.5 \times 10^3 \times 1000}{0.5}$ = 5 scm² mol-1 = 5 ×10 4 Sm2 mol-1 (b) 0.13: K= 2:303 log 100-99 Ans: 0.693 = 2.303 lover 100 6.93 t I by solving It = 46.06 min Q.14 : (C) Acidic strength & (+)ve oxidation No. Ans.

0.15 (9) Ans: Linkage isomerism shown by complexy having ambidentale ligands. Q.16 (b) Ans magnetic moment & No of unpaired electron as M= TN(n+2) B.M. on p > Fer > 3d5 = 5 unpaired E m Q > V+2 > 3d 3 > 3 unpaired c m R ⇒ Fe2+ ⇒ 3d 6 ⇒ A impaired E more will be stability of antocation higher will be the stability of alkye halide C.17. (a) Bry towards SN+ Receipt. 1e 3°72°710 By CH3 COOH Lifterna > CH3 CH2OH PUS > CH3-CH2-CE Alc KOH CH2=CH2 € eshylene. Q.19 (b) ni>I>II 7 III - lore pair free I - Netrogen is sp2 hybridized Ans I - love pair in Resonance. Q.20 (b) Ars: It is due to different configuration at CI carbon Q.21 Ans= 10. 0 0 oris Ð CHZ CHZ (10) cn 3 Q. 22) 9 Ans. maximum No. of electron 2n2 = 2232 = 18 . No. 7 election with ms = - 1 will be 9

Q-23 > 4 H B H B H Ary . B2H6 N = N = 0 ~ N = N - 0 ~ N = N = 0 NO2 NOOA >0 H2 5203 01 10 H2 5208 H-NA B3 H316 S. And het the salubility of Agel 13 'se mal liter and that of Gill is ymale lides! Agel = Agt + LE cut + ce y y Cyle ksp of Agu = [Ag+] [LE] 1.6×10" = x(x+y) -0 Similarly, Ksp of cull = [Cut] [CU] 1.6×10 = Y(k+y) -1 on solving eq. 0 \$ (1) [Ag+] = 1.6×107 · [X=7] Q.25 7 8 CH3 CH3-CH2-CH2-CH2 le Ensentioneric poir =2 Ang н снз CH3-CH2 - C* CH-CH3 Chanto meric pair = 4 4 10 CH3-CH2- C- CH2-CH3 ce $CH_{3}-CH_{2}-CH_{2}-CH_{3}=1$

Math

$$= \sum_{a} \int_{a}^{a} \int_{a}^{a}$$

$$S = \sum_{s=0}^{20} 30_{c_{s}} Sin(60-3)x (20(52) - 6)$$

And $8_{0}n(0) e(2)$

 $S = \sum_{s=0}^{20} 30_{c_{s}} (3in(5x) (20(50-5)x + 8in(60-5)x(200(5x)))))$

 $= \frac{3}{2}0 30_{c_{s}} Sin(302)$

 $S = 0 30_{c_{s}} Sin(30$

$\frac{12}{12}$ $(ffred t)^3 = 3 = 2h$ and $24 + 2 = 2k$
(K-1)2= 24+3 , t=K-1
$(h,k) \rightarrow (\lambda \rightarrow \gamma) = (\chi + 3)$
$(-3, \pi^2)$ B $P[t^2, 2t]$
$2+3_{2}=-2_{4}=3+2=0$
$\frac{1}{2} = \frac{1}{a_{-1}} + \frac{dy}{dt} = 1 \qquad \qquad y \frac{(y'+1)}{2y'} + \frac{x(y'+1)}{-2} = 1$
$\Rightarrow a = \frac{dy}{dx^{-1}} \qquad \Rightarrow (y'+1)(y-xy') = 2y'$
$a_{t1} = \underbrace{2d_{y}}_{dx} e a_{-1} = -\underbrace{2}_{dy}$ $\underbrace{d_{x}}_{dy + 1} e a_{t1} = -\underbrace{2}_{dy}$ $\underbrace{d_{y}}_{dy + 1} d_{y} = \underbrace{d_{y}}_{dy}$
$= \sum_{k=0}^{n} + aut \left(\frac{(k+y)}{1+(k+y)} \frac{2}{2} - \frac{(k+y)}{2} \frac{2}{2} \right)$ $= \frac{1}{1+(k+y)} \frac{2}{2} \cdot \frac{(k+y)}{2} \frac{2}{2}$
Su = 2 tant (8+4) 2 - tant (8+3) 22
Su = tait (1+1)3/2 - tait 32/2
$J_{a0} = \overline{D_{g}} - tant_{gr} = 1$
$\begin{array}{c} 1 \\ 1 \\ 1 \\ 2 \\ 2 \\ 2 \\ 2 \\ 2 \\ 2 \\ 2 \\$
$\chi = \frac{2}{3} \cot 1$
些 雨
$\frac{15 \cdot 1}{50} \frac{2 S_{(3)} + 2 S_{(2)}}{50 C_{2} + 50 C_{2}} = \frac{12}{49}$
+ >0

$$\frac{24}{150} \frac{1}{150} \frac{(1+x+x^{2}+x^{2})^{10}}{(1+x+x^{2}+x^{2})^{10}} = a_{0}+a_{1}x+\dots+a_{300}x^{300}} \frac{a_{00}}{a_{0}}x^{200}}{a_{0}} \frac{a_{1}}{b_{0}} \frac{b_{1}b_{1}b_{1}b_{2}b_{2}b_{2}b_{2}b_{2}b_{2}}{a_{1}} = a_{1}+2a_{2}+3a_{3}+\dots+3a0}a_{3a0}x^{2a0}}{a_{2a0}-6}$$

$$\frac{b_{1}}{b_{1}}\frac{x-1}{a_{1}} \frac{1}{a_{1}}a_{0} = a_{0}+a_{1}+a_{2}+\dots+a_{3m}-6}{a_{3m}} - \frac{a_{1}}{a_{3m}} - \frac{a_{1}}{a_{2m}} -$$